

8.23 Determination of Stefan's Constant (Laboratory Method)

The laboratory apparatus used to determine the Stefan's constant is shown in Fig. 8.12. A hollow hemispherical metallic vessel A is enclosed in a wooden box W . The inner surface of A is coated with

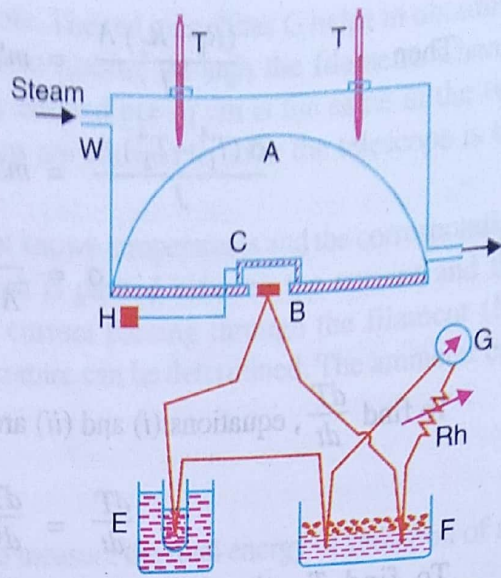


Fig. 8.12

The actual experiment consists of two parts.

1. The thermocouple is first standardized. Before passing steam into the chamber, the disc B is at the room temperature. The water bath E acts as a hot junction. It is covered and at various temperatures of the hot junction, the corresponding deflections in the galvanometer are noted. A graph between the difference of temperature of the hot junction and the room temperature along the Y-axis and galvanometer deflection along X-axis is plotted (Fig. 8.13). From the graph,

$$\frac{dT}{d\theta} = \tan \alpha = \frac{AB}{BC} \dots(i)$$

2. The disc is completely covered with C and steam is passed into the chamber. After some time, the thermometers TT show constant temperature. The bath E is at room temperature. With the help of the handle H, the cover C is tilted so that the upper surface of the disc B receives the radiations from the enclosure. The deflections in the galvanometer are observed after equal intervals of time (say 10 seconds). A graph is plotted between time and deflection (Fig. 8.14). A tangent is drawn on the curve at a point D.

$$\frac{dt}{d\theta} = \tan \beta = \frac{EF}{GF} \dots(ii)$$

Let, at any instant, the temperature of the enclosure and the disc be T_1 and T_2 (degrees Kelvin) respectively. The disc will absorb more heat from the surroundings and radiate less heat to the surroundings. Its temperature will rise. From Stefan's law,

$$R_1 = \sigma T_1^4 \text{ and } R_2 = \sigma T_2^4 \dots(iii)$$

$$(R_1 - R_2) = \sigma (T_1^4 - T_2^4)$$

Here, R_1 is the amount of heat radiation absorbed per unit area per second by the disc and R_2 is the amount of heat radiation emitted per unit area per second by the disc. Let the mass of the disc be m , specific heat S , rate of rise of temperature dT/dt , and area of the surface A .

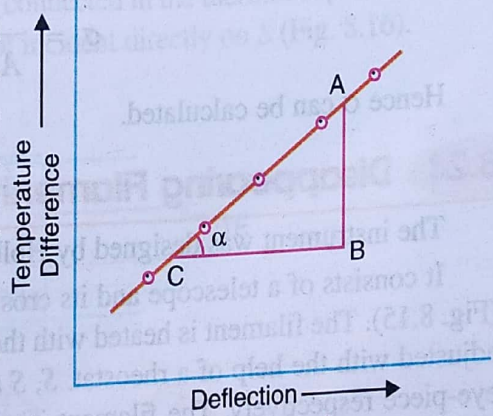


Fig. 8.13

Then

$$\frac{(R_1 - R_2) A}{J} = mS \frac{dT}{dt}$$

$$\sigma \frac{(T_1^4 - T_2^4) A}{J} = mS \frac{dT}{dt}$$

$$\sigma = \frac{JmS}{A(T_1^4 - T_2^4)} \times \frac{dT}{dt} \dots (iv)$$

To find $\frac{dT}{dt}$, equations (i) and (ii) are used.

$$\frac{dT}{dt} = \frac{dT}{d\theta} \times \frac{d\theta}{dt} = \frac{\tan \alpha}{\tan \beta}$$

To find T_2 , the deflection in the galvanometer corresponding to the point *D* on graph in Fig. 8.14 is noted and for this deflection, the temperature difference from the graph (Fig. 8.13) is noted. To this reading add the room temperature and find T_2 in degrees Kelvin.

Substituting these values in equation (iv)

$$\sigma = \frac{JmS \tan \alpha}{A(T_1^4 - T_2^4) \tan \beta}$$

Hence σ can be calculated.

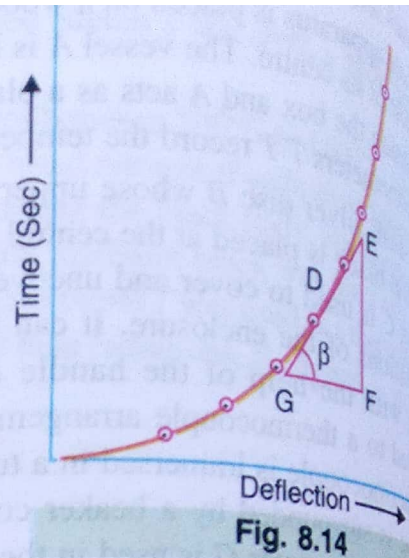


Fig. 8.14

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